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6

7 **Systematic Review of Dairy Processing Sludge and Secondary** 8 **STRUBIAS Products Used in Agriculture**

9 **Yihuai Hu¹, Olha Khomenko^{2,5}, Wenxuan Shi^{2,3}, Ángel Velasco-Sánchez^{4,8}, S.M.**
10 **Ashekuzzaman², Nadia Bennegadi-Laurent⁴, Karen Daly², Owen Fenton^{*2}, Mark G. Healy³,**
11 **J.J. Leahy⁵, Peter Sørensen⁶, Sven G Sommer¹, Arezoo Taghizadeh-Toosi⁷, Isabelle**
12 **Trinsoutrot-Gattin⁴**

13 ¹Department of Biological and Chemical Engineering, Aarhus University, Finlandsgade 12, 8200
14 Aarhus N, Denmark, ²Teagasc, Johnstown Castle, Environment Research Centre, Wexford,
15 Ireland, ³Civil Engineering and Ryan Institute, College of Science and Engineering, National
16 University of Ireland, Galway, Ireland, ⁴UniLaSalle, Aghyle, Rouen, Rue du Tronquet, 76130
17 Mont-Saint-Aignan, France, ⁵Department of Chemical Sciences, School of Natural Sciences,
18 University of Limerick, Plassey Park, Limerick, Ireland, ⁶Department of Agroecology, Aarhus
19 University, Blichers Allé 20, 8830 Tjele, Denmark, ⁷Danish Technological Institute, DTI, Agro
20 Food Park 15, Skejby, 8200 Aarhus N, Denmark, ⁸Soil Biology Group, Wageningen University
21 and Research, Droevendaalsesteeg 3, 6700 AA Wageningen, The Netherlands.

22 ***Correspondence:**

23 Corresponding Author

24 owen.fenton@teagasc.ie

25

26 Worldwide dairy processing plants produce high volumes of dairy processing sludge (DPS), which
27 can be converted into secondary derivatives such as struvite, biochar and ash (collectively termed
28 STRUBIAS). All of these products have high fertilizer equivalent values (FEV), but future
29 certification as phosphorus (P)-fertilizers in the European Union will mean they need to adhere to
30 new technical regulations for fertilizing materials i.e., content limits pertaining to heavy metals
31 (Cd, Cu, Hg, Ni, Pb and Zn), synthetic organic compounds and pathogens. This systematic review
32 presents the current state of knowledge about these bio-based fertilizers and identifies knowledge
33 gaps. In addition, a review and calculation of greenhouse gas emissions from a range of concept
34 dairy sludge management and production systems for STRUBIAS products (i.e. biochar from
35 pyrolysis and hydrochar from hydrothermal carbonization (HTC)) is presented. Results from the
36 initial review showed that DPS composition depends on product type and treatment processes at a
37 given processing plant, which leads to varied nutrient, heavy metal and carbon contents. These
38 products are all typically high in nutrients and carbon, but low in heavy metals. Further work needs
39 to concentrate on examining their pathogenic microorganism and emerging contaminant contents,
40 in addition to conducting an economic assessment of production and end-user costs related to
41 chemical fertilizer equivalents. With respect to STRUBIAS products, contaminants not present in
42 the raw DPS may need further treatment before being land applied in agriculture e.g., heated
43 producing ashes, hydrochar or biochar. An examination of these products from an environmental
44 perspective shows that their water quality footprint could be minimised using application rates
45 based on P incorporation of these products into nutrient management planning and application by
46 incorporation into the soil. Results from the concept system showed that elimination of methane
47 emissions was possible, along with a reduction in nitrous oxide. Less carbon (C) is transferred to
48 agricultural fields where DPS is processed into biochar and hydrochar, but due to high

49 recalcitrance, the C in this form is retained much longer in the soil, and therefore STRUBIAS
50 products represent a more stable and long-term option to increase soil C stocks and sequestration.

51

52

53 **Keywords:** Circular economy, bio-fertilizer, phosphorus, environment-agriculture, gaseous
54 emissions

55

56 **Abbreviation List**

57 Ammonia (NH₃)

58 Anaerobic digesters (AD)

59 Biological Oxygen Demand (BOD)

60 Carbon (C)

61 Carbon dioxide (CO₂)

62 Cheese Whey Wastewater (CWW)

63 Chemical Oxygen Demand (COD)

64 Colony forming units (CFU)

65 Community-level physiological profiling (CLPP)

66 Completely Stirred Tank Reactor (CSTR)

67 Dairy Processing Sludge (DPS)

68 Dissolved air floatation (DAF)

69 Dry Matter (DM)

70 European Union (EU)

71 Fertilizer equivalent value (FEV)

72 Greenhouse gas (GHG)

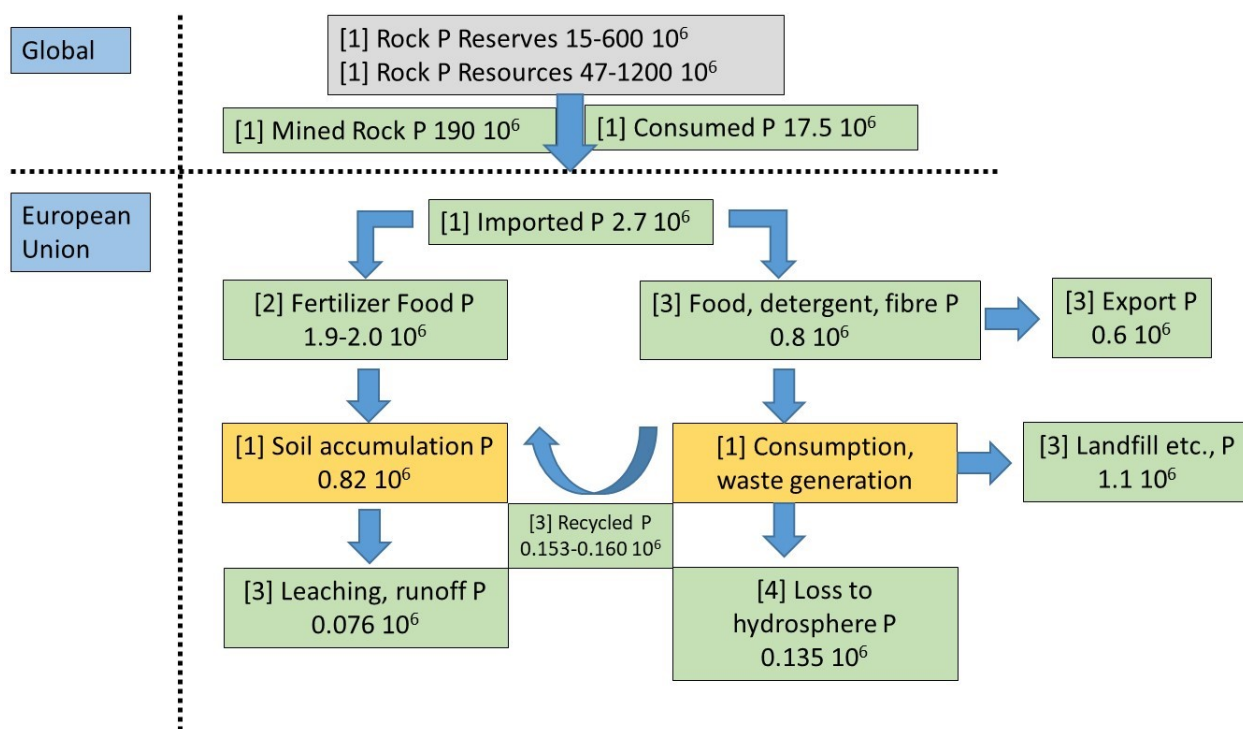
73 Hydrothermal Carbonification (HTC)

- 74 Integrated Farm Systems Model (IFSM)
- 75 Mean residence time (MRT)
- 76 Membrane Anaerobic Reactor System (MARS)
- 77 Membrane Reactor (MBR)
- 78 Methane (CH₄)
- 79 Mineral Fertilizer Equivalence (MFE)
- 80 Mineral Fertilizer Replacement Value (MFRV)
- 81 Nitrogen (N)
- 82 Nitrous oxide (N₂O)
- 83 Organic Matter (OM)
- 84 Organic Trace Compounds (OTCs)
- 85 Oxygen (O₂)
- 86 Phosphorus (P)
- 87 Phospholipid fatty acid assays (PLFAs)
- 88 PolyChloroBiphenyls (PCBs)
- 89 Polycyclic Aromatic Hydrocarbons (PAHs)
- 90 P solubilising microorganisms (PSM)
- 91 Pyrolysis (PC)
- 92 Rotating Biological Contractors (RBC)
- 93 Sequencing Batch Reactor (SBR)
- 94 STRUBIAS (struvite, biochar, ashes)
- 95 Total ammonium nitrogen (TAN)
- 96 Up-flow Anaerobic Sludge Blanket (UASB)
- 97 Volatile Solids (VS)
- 98

99 INTRODUCTION

100 Mineral phosphorus (P) is a listed European Union (EU) critical raw material due to its importance
 101 in food production (European Commission, 2017; Espinosa et al., 2020). As agriculture is the
 102 largest consumer of mined P in Europe (1.1 million tonnes in 2015; Eurostat, 2020), security of
 103 supply may be challenging because the source of non-renewable rock phosphate is in geopolitically
 104 sensitive regions. The dairy processing sector produces P-rich dairy processing sludge (DPS)
 105 which, when used directly or in derived secondary products such as STRUBIAS (STRUvite,
 106 BIOchar, AShes), may reduce the dependence on mined rock P (Figure. 1). The European dairy
 107 processing industry processed about 144.6 million tonnes or 140.4 billion litres of domestic milk
 108 in 2020 (Table 1), about 46% more than the USA which is the second largest milk producing
 109 country in the world (Agriland, 2020). It is estimated that dairy food processing wastewater
 110 treatment can generate up to 20 kg (mean 17.45 kg m⁻³) DPS per m³ of milk processed
 111 (Ashekuzzaman et al., 2009a), which resulted in 2.45 million tonnes of DPS (wet weight) across
 112 the EU in 2020 (Table 1).

113



114

115 **Figure 1.** Global P resources and consumption of P and associated P usage streams in the EU in
 116 tons per year. Based on Ott and Rechenberger (2012), Scholz and Wellmer (2013) and Schoumans
 117 et al. (2015).

118

119 **TABLE 1** | EU domestic milk intake and associated estimated dairy processing sludge
 120 generation (wet weight) with total phosphorus and nitrogen quantity.

Year	2019	2020
Milk Production (million tonnes) ¹	142.8	144.6
Dairy processing sludge (million tonnes) ²	2.42	2.45
Total Phosphorus dairy sludge (tonnes) ³	12680	12840
Total Nitrogen dairy sludge (tonnes) ³	17272	17490

121 ¹ Data source: CSO (2021a,b); ² Estimated sludge to raw milk ration (kg m^{-3}) of 17.45
 122 was used from Ashekuzzaman et al. (2019a) to estimate sludge generation and
 123 density of milk 1030 kg m^{-3} (at 20 °C) as per Walstra et al. (2005) was used for mass
 124 to volume conversion; ³ Median P and N concentration of 35.9 and 48.9 g kg^{-1} (dry
 125 weight) and median dry matter content of 14.6% were used estimate total P and N
 126 content, respectively, data source: Ashekuzzaman et al. (2019a).

127

128 Phosphorus in DPS can be recycled to fields, creating a circular economy, and can be used as a
 129 replacement for mineral P fertilizer produced from mined rock phosphate (Mayer et al., 2016).

130 Similar to other organic fertilizers, land application only occurs in growing seasons (Sommer and

131 Knudsen, 2021), meaning that storage is required for extended periods, which increases the cost

132 of management. Storage and land application of DPS is a source of the greenhouses gases (GHGs),

133 methane (CH_4) and nitrous oxide (N_2O) (Smith et al., 2021), and when applied to soil may result

134 in incidental losses of nutrients along surface or near surface pathways (Fenton et al., 2017; Shi et

135 al., 2021a). As production becomes more centralised on a smaller number of larger farms in some

136 EU member states (e.g. Denmark or France), the land bank opportunities for DPS application may

137 become limited depending on the land use change and farmer willingness to accept DPS. Due to

138 the nutrient content value and related transport costs, a land bank radius of about 10 km around a

139 processing plant pertains. This may cause an oversupply to land areas near the processing plant
140 and an increased risk of P losses to water (i.e. critical source areas). In addition to centralization
141 of production, future application rates of fertilizer will be limited by P application, and not by
142 nitrogen (N) rates. Presently, the limits imposed on N application rates have led to significant over-
143 application of P (i.e. P:N ratio higher than the ratio needed by crops) and increased potential for
144 eutrophication in regions with a high proportion of sludge and slurry production (Lu et al., 2012).
145 As an example, in Brittany, France, the issue of P and eutrophication is of major importance due
146 to the local high permeability bedrock (granite, shale and sandstone) and agricultural (high level
147 of livestock density) context, which prevents localised organic fertilizer land application. This
148 means there is not always a match between DPS and suitable land availability (e.g. P-deficient
149 soils) in the local area near the dairy processing plant (Le Noë et al., 2018). Such conditions have
150 created a market for the transport and application of DPS e.g., the total cost of spreading DPS
151 (including transport, analyses, spreading and monitoring) is between 20 and 30 € per tonne in
152 France (Laperche, 2014). Any new regulations pertaining to the dairy industry will incentivise
153 producers to reduce the volume and weight of DPS transported over long distances to end users.
154 The post processing of DPS to produce secondary STRUBIAS products is a potential solution in
155 this regard.

156 The main objective of this systematic review is to collate information that will help give DPS and
157 their secondary products certification as P-fertilizers in accordance with technical proposals for
158 new fertilizing materials under the Fertilizing Products Regulation (European Commission, 2003,
159 2019; Huygens et al., 2019). The purpose of this systematic review is to contribute to decision
160 making about the most sustainable transformation of these “wastes” into high value bio-fertilizers
161 used by both traditional and organic farmers.

162

163 **MATERIALS AND METHODS**

164 **Search criteria**

165 All search results were evaluated using the PRISMA statement (Page et al., 2021). In a systematic
166 review, information was collated to answer the following two research questions: (1) how DPS
167 and STRUBIAS production on *in-situ* and *ex-situ* treatment and processing units affect product
168 characteristic, fertilizer replacement value (FEV) and P dynamics in soil, risks of GHG emissions
169 and pathogen and heavy metal pollution? (2) how to combine this information to develop
170 production systems necessary to certify DPS and STRUBIAS products as P-fertilizers in
171 accordance with technical proposals for new fertilizing materials under the Fertilizing Products
172 Regulation? A comprehensive systematic literature search of three online databases was
173 performed, Scopus (www.scopus.com), PubMed (<https://pubmed.ncbi.nlm.nih.gov/>) and Web of
174 Science (<https://apps.webofknowledge.com>). Searches were conducted in English on literature
175 from 1983 to (1 July) 2021. All research articles related to DPS and STRUBIAS were identified.
176 In addition, Google Scholar was used to find reports pertaining to some sections of this review.
177 The search terms and keywords used to identify research studies were: dairy processing
178 sludge/waste, fertilizer replacement value, phosphorus, circular economy, treatment,
179 characterization, composition, heavy metals, greenhouse gas, methane, nitrous oxide, *E. coli*, and
180 PAHs (Polycyclic Aromatic Hydrocarbons). Studies that did not contain an abstract in English
181 were excluded from this study during the screening stage. As there were not enough published
182 studies dealing with DPS and STRUBIAS, no meta-analysis was possible for this paper.

183

184 **Screening of Search Results**

185 Duplicates were removed manually and abstracts were screened by two screeners against the target
186 research questions. Exclusion criteria, described below, were developed and selected and cross-
187 checking was performed on these excluded articles. If disagreements between the two screeners
188 occurred, a third screener adjudicated. Full-text review was independently conducted by three
189 reviewers and reasons for exclusion were annotated and tracked (e.g. “data pertaining to a different
190 waste other than DPS or STRUBIAS”). The primary reasons for excluding papers were: (i) articles
191 completely un-related to search questions; (ii) general knowledge papers; and (iii) papers that did
192 not follow the basic criteria of scientific research (e.g. experimental design with sufficient
193 replication). Articles clearly meeting the inclusion criteria were obtained for full-text review unless
194 unavailable. These included articles related to the search inquiry, providing that scientific
195 laboratory experiments or field studies had a minimum number of replicas and a negative control.
196 Articles were not considered further when their title and abstract clearly indicated that the study
197 did not meet the inclusion criteria (see Supplementary Figure 1). The studies included in tables
198 consist of those that were reviewed in detail.

199

200 **Greenhouse Gas Emission**

201 Some data needed for the holistic review were not available in the literature and needed to be
202 developed outside of the systematic review within the present study. Greenhouse gas emission
203 from DPS sludge and secondary STRUBIAS products was calculated using a whole systems
204 approach. Herein, methods presented calculate CH_4 , N_2O and ammonia (NH_3) emissions from
205 sludge and secondary STRUBIAS products from production to field application. Ammonia
206 emission is included in the calculation due to risk of N_2O emission from NH_3 deposition to land.
207 In this analysis, various conceptual scenarios (from production to field application) are compared
208 with the aim to find potential scenarios that can minimise or eliminate emissions.

209 Assumptions for management of sludge and STRUBIAS products for a Danish dairy production
210 site are used in the scenarios as follows: each month 1/12 of one ton of annual produced standard
211 sludge is transferred to a store and the temperature in the stored sludge is similar to the monthly
212 average air temperature of Denmark. The tanks with stored sludge are emptied at the start of April
213 and subsequently land applied to fields in April. In the scenarios where sludge is processed to

214 biochar or hydrochar, the same monthly amount is treated and the products are land applied in
215 April. The annual amount of DPS treated is 1 ton.

216 In the calculations a one-hundred-year global warming potential (GWP100) for CH₄ were set to
217 34 kg CO₂eqv kg⁻¹[CH₄] and N₂O to 298 kg CO₂eqv kg⁻¹[N₂O] (Myhre et al., 2013).

218

219 **Calculation of CH₄ emission**

220 Methane emissions from stored liquid sludge is calculated with the CH₄ emission model as used
221 in the Integrated Farm Systems Model (IFSM) for manure management in beef and dairy
222 production systems (Chianese et al., 2009). This has been used previously to assess the impact of
223 GHG reduction strategies in agriculture (Rotz and Hafner, 2011; Dutreuil et al., 2014). Equation
224 1 is used to calculate CH₄ emission from anaerobic stored livestock liquid manure and digestate
225 from biogas plants if concentration volatile solids (VS) and air temperature is known (Baral et al.,
226 2018):

$$227 \quad F_{CH_4} = VS_D \times (e^{\left[\ln A - \frac{E_a}{RT}\right]}) \times 24 \quad (1)$$

228 where F_{CH₄} is CH₄ emission rate (g CH₄ kg⁻¹ VS day⁻¹), A is the pre-exponential factor of 31.2 g
229 CH₄ kg VS⁻¹ h⁻¹ (Petersen et al., 2016), E_a the apparent activation energy set to 81 kJ mol⁻¹
230 (Elsgaard et al., 2016) giving the temperature response of CH₄ production. R is the gas constant
231 (8.314 J mol⁻¹ K⁻¹) and T the temperature (K). The degradable fraction V_{SD} is 56% of VS in the
232 sludge (Smith et al. 2021). This equation and the parameters presented here is used to calculate
233 CH₄ emission from stored DPS.

234

235 **Emission of NH₃ and N₂O**

236 In the Danish GHG emission inventory (Nielsen et al., 2018), it is assumed that 0.5% of total N is
237 emitted in form of N₂O from stored liquid manure and sludge, and 1% of total-N is emitted from
238 slurry and sludge applied to soil (based on the standard emission factors given by IPCC (2019)).
239 Another assumption is that N₂O emissions from soil N are unaffected by any application of biochar

240 or hydrochar applied to soil. When struvite is applied, it is assumed that 1% of the ammonium
 241 (NH_4^+) is emitted as N_2O as is the case for mineral fertilizer N applied to soil (IPCC, 2019)

242 During storage of the sludge 34% of the proteins are transformed to total ammonium nitrogen
 243 ($\text{TAN}=\text{NH}_3 + \text{NH}_4^+$) (Mottet et al., 2010) and may be emitted from the storage. Emission of NH_3
 244 from stores with a cover of PVC roof are set to 2.6% of TAN i.e. similar to emissions from covered
 245 stored liquid manure (Hansen et al., 2008) and of sludge injected into black soil to 2% of TAN
 246 (Olesen et al., 2020). The negative charge of biochar will contribute to a negligible NH_3 emission
 247 from soils to which hydro-biochar is applied (Chu et al., 2019), biochar from pyrolysis do not
 248 contain TAN and the NH_4^+ in struvite is assumed not to volatilize due to the low pH (Sommer et
 249 al., 2004). At deposition on land or water, a fraction of NH_3 will be transformed to N_2O and be
 250 emitted to the atmosphere. This indirect N_2O emission is estimated using Equation 2 (IPCC, 2019):

$$251 \quad F_{\text{N}_2\text{O}} = F_{\text{NH}_3} * 0.01 * \frac{44}{28} \quad (2)$$

252 where $F_{\text{N}_2\text{O}}$ is given in kg N_2O , F_{NH_3} is given in kg $\text{NH}_3\text{-N}$ emitted, 0.01 is a default factor given
 253 by the IPCC for calculation of the climate warming effect of emitted NH_3 , and 44/28 is to calculate
 254 from concentration given in $2*\text{N g mol}^{-1}$ to $2*\text{N}+\text{O g mol}^{-1}$ (IPCC, 2006).

255

256 **Carbon sequestration**

257 During storage of untreated sludge, a fraction of C is lost in form of carbon dioxide (CO_2) and
 258 CH_4 . This emission is calculated assuming that 2.6 kg VS is lost in the form of CO_2 and CH_4 for
 259 each kg of CH_4 produced. The sludge is heated in hydrothermal carbonization due to the oxidation
 260 of C, and in this process about 30 % of C is lost in the form of CO_2 , and about 70% of C in treated
 261 biomass is retained in the hydrochar (Kambo and Dutta, 2015). In the pyrolysis process, more CO_2
 262 is lost and between 25-35% (avg. 30%) of solids (40% C) are retained in the biochar (Kambo and
 263 Dutta 2015).

264 In the calculations used herein, the C retention of field applied sludge-C was set to 25%, which is
 265 between 12% of manure C input in the longer term (avg. 18 years; Maillard and Angers, 2014) and
 266 35% for a 20-year period by scaling results from a study of transformation of C in digestate applied

267 to soil (Thomsen et al., 2013). The longer retention time of C in sludge than in animal slurry was
268 due to sludge organic matter (OM) from a wastewater treatment plant that has been transformed
269 to a more stable form of C. It is calculated that C concentration in VS is 517 g C kg VS⁻¹, and this
270 estimate is used to calculate C concentration in sludge, where OM is measured as VS.

271 It has been calculated that between 90% and 97% of the C in biochar from pyrolysis of a range of
272 different biomasses will still remain in the soil after 100 years (Lehmann and Joseph, 2015; Wang
273 et al., 2016). No study of the recalcitrance of biochar from pyrolysis of dairy sludge was found in
274 the literature, and a conservative/cautious estimate is that 90% of the C in biochar from pyrolysis
275 (PC) of dairy sludge is recalcitrant.

276 Hydrochar from hydrothermal carbonization (HTC) is produced at a lower temperature than when
277 producing biochar by pyrolysis, and C component in the hydrochar has been oxidised less than
278 pyrolysis products, i.e. less oxygen is added during the process. Our assumption is that C in
279 hydrochar is less recalcitrant than biochar produced by pyrolysing dried biomass, and that only
280 50% C in hydrochar is recalcitrant. This assumption is supported by the study of Malghani et al.
281 (2013), who stated that “although both HTC and PC chars were produced from the same feedstock,
282 PC chars had markedly higher potential for carbon sequestration than HTC”.

283

284 **RESULTS AND DISCUSSION**

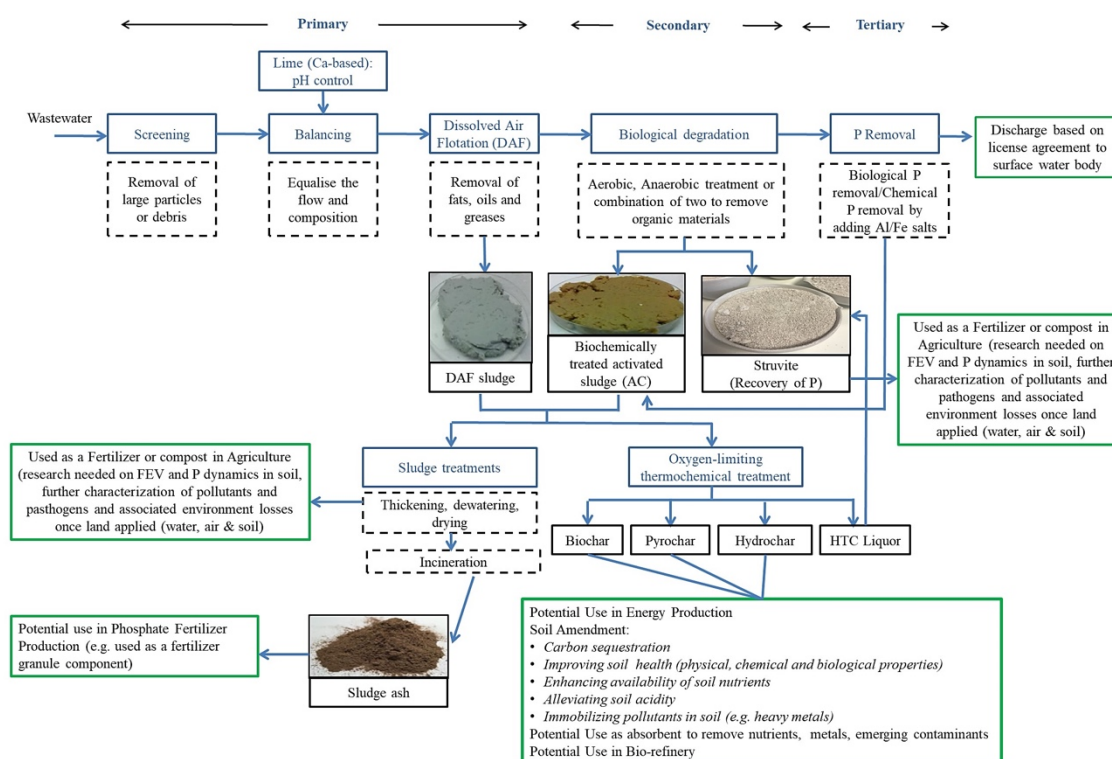
285 **Treatment Options at Dairy Processing Plants, and Volumes and Composition of DPS**

286 The dairy processing industry generates a large volume of waste, which is high in OM. Discharge
287 licencing, which is site specific, aims to prevent a reduction in surface water oxygen (O₂)
288 concentrations and the onset of surface water eutrophication (Neal and Heathwaite, 2005). To
289 achieve such discharge thresholds, a chain of treatment before discharge at dairy processing plants
290 is needed (Figure 2). In fact, most DPS applied to land in the EU is first treated, composted or
291 incinerated before being applied to land (Figure 2). Research now focusses on converting DPS into
292 secondary STRUBIAS products (Figure 2).

293

294 Dairy Biological Wastewater Treatment

295 A schematic of typical dairy wastewater treatment processes and DPS generation is shown in
 296 Figure 2. Pre-treatment removes particulate matter and retains fats that could interfere with
 297 subsequent treatment. The pre-treatments and physio-chemical treatments most frequently used on
 298 dairy processing plants are buffer tanks, neutralization, sieving, flotation and degreasing (Droste
 299 and Gehr, 2018). Flocculation (coagulation) is the most simple and economical pre-treatment
 300 method, and reduces water turbidity by reducing particulate substances and fats, which could
 301 interfere with subsequent treatments (Carvalho et al., 2013). Pre-treatment by flocculation
 302 enhanced by the presence of lactic acid bacteria (*Lactobacillus plantarum*), which ferment the
 303 lactose and produce lactic acid. The acidity of this compound precipitates the milk proteins, and
 304 thus significantly reduces the chemical oxygen demand (COD). This reduction increases with
 305 addition of flocculants such as chitosan or carboxymethyl cellulose (CMC) to this solution. The
 306 COD removed varied by 65 to 78% for CMC and 49 to 82% for chitosan (Dyrset et al., 1999).



307

308 **Figure 2.** Dairy wastewater treatment flowcharts showing DPS and STRUBIAS products
 309 (modified from Shi et al. (2021a)).

310 In the secondary treatment, sludge undergoes aerobic treatment that is cost efficient and
311 controllable (Kolev Slavov, 2017). This treatment step can reduce more than 90% of COD (Carta-
312 Escobar et al., 2004; Kushwaha et al., 2011; Carvalho et al., 2013). In aeration ponds, the aerobic
313 pathway uses microorganisms contained in biological reactors, which in the presence of O₂,
314 degrade the OM suspended in the sludge.

315 The soluble organic components transformed and emitted mainly in the form of CO₂ and NH₃,
316 while the insoluble pollution fraction, including the microorganisms, is recovered by separation
317 and form "activated" sludge (Sustarsic, 2009). A large number of aerobic treatment types are being
318 used e.g., rotating biological contractors (RBCs), sequencing batch reactors (SBRs), or membrane
319 reactors (MBRs) (Goli et al., 2019). In particular, SBRs are effective at varying loading capacities
320 (Kolev Slavov, 2017). In this step, a single tank used for filling, aeration, settlement and effluent
321 withdrawal, and recycling of solids. The sludge produced during primary and secondary treatment
322 may be treated anaerobically in biogas reactors. During this process, OM is transformed by
323 microorganisms to CH₄ and CO₂. This process produces small volumes of sludge (less than 0.05
324 kg of dry matter (DM) per kg of COD eliminated) (Omil et al., 2003). The most common anaerobic
325 reactors are up-flow anaerobic sludge blanket (UASB), completely stirred tank reactors (CSTRs)
326 and membrane anaerobic reactor systems (MARSs). Finally, where needed, the N, P,
327 micropollutants or pathogenic microorganisms in the wastewater can be further reduced using a
328 finishing treatment e.g. for P content reduction alternation between aerobic/anaerobic conditions
329 or P coagulation and precipitation by ferric chloride or aluminum sulphate can be deployed (Rivas
330 et al., 2010). After these treatments, the particles settle in a clarification pond. The supernatant is
331 discharged, while the precipitated sludge is dewatered after addition of polymers to promote
332 flocculation and separation between water and suspended matter (dewatering). This is then

333 thickened (thickening) during its passage through dewatering grids. This sludge is stored in tanks
334 or treated further and then used as a bio-based fertilizer (Huygens et al., 2019).

335

336 **Wastewater Volumes and Contents**

337 The volume of wastewater produced at processing plants can be high and is product-dependent.
338 For example, one litre of processed milk can produce up to 10 L of effluent (Lateef et al., 2013).
339 In the cheese manufacturing industry, the whey is the main pollutant discharged to water and soil.
340 This is mainly due to its high carbohydrate content (4-5%) of which lactose is the main constituent
341 (Kolev Slavov, 2017). Consequently, DPS produced from cheese production, called Cheese Whey
342 Wastewater (CWW), contains high concentrations of organic components contributing to a high
343 COD and biological oxygen demand (BOD) (Ahmad et al., 2019). The COD of CWW is higher
344 ($0.79\text{-}77.3\text{ g L}^{-1}$) than COD of milk plant effluent ($0.183\text{-}10\text{ g L}^{-1}$) (Carvalho et al., 2013) or milk-
345 treated condensate wastewater ($<0.001\text{ g L}^{-1}$), which is the water obtained during the concentration
346 and evaporation processes of milk and its by-products (Bourbon and Huet, 2018). The CWW
347 contains some milk or milk by-products, oils and greases, and cleaning water containing sterilising
348 agents, acid and alkaline detergent (Carvalho et al., 2013; Ahmad et al., 2019). As a result, the
349 concentration of inorganic compound may be heterogeneous, with ranges observed for P from 8
350 to 510 mg L^{-1} and N from 14 to 1462 mg L^{-1} (Demirel et al., 2005).

351 The amount and composition of DPS is determined by the composition of milk and use of
352 additives, the dairy production line and cleaning and disinfectant products used for cleaning. In
353 addition, wastewater treatment, varies between plants (Figure 2), and depends on the type of dairy
354 processing involved (Rico Gutiérrez et al., 1991; Karadag et al., 2015; Ahmad et al., 2019; Shi et
355 al., 2021a). Recently, physical and chemical characteristics of 63 DPS samples (9 dairy processing
356 plants in Ireland) were reported (Ashekuzzaman et al., 2019a). The main DPS types included in
357 the study were bio-chemically treated activated sludge leading to aluminium-precipitated sludge
358 (Al-DPS) and iron-precipitated sludge (Fe-DPS) depending the dosing of alum or ferric salt to
359 remove P, and to lime-stabilised calcium-precipitated sludge (Ca-DPS) generated after dissolved
360 air floatation (DAF). In some processing plants, mixing of DPS generated from aerobic biological

361 wastewater treatment and DAF processes occurs before land application or further disposal
362 processing. A few of the examined plants have anaerobic digesters (AD) that produced AD sludge.
363 That concentration of components varies between sludge categories and treatments and differences
364 in composition was highest for N, P and K (Ashekuzzaman et al., 2019a). This difference is higher
365 between activated and DAF sludge types than those in the combined sludge (sludge mixed from
366 both activated and DAF process). The addition of chemicals such as Fe-, Al- or Ca-based
367 coagulants during the removal of P from the effluent may explain these results. In addition, the N
368 concentration was lower in DAF sludge compared to AD sludge in contrast to the increased P
369 concentration (Ashekuzzaman et al., 2019a). The addition of lime during the DAF process explains
370 this result by an increase of the pH, which promotes the volatilization of N in the form of NH_3 . On
371 the other hand, anaerobic degradation during the AD process increases the TAN concentration in
372 the sludge due to organic N mineralization.

373 Tables 2 and 3 illustrate the physicochemical characteristics of dairy processing effluents and DPS
374 reported in the literature. These characteristics highlight the variability in dairy effluent
375 compositions related to the type of bio-products processed (i.e. milk, cheese, yogurt and butter)
376 (Omil et al., 2003; Carvalho et al., 2013; Karadag et al., 2015), and the wastewater treatment
377 processes used (Britz et al., 2006; Ashekuzzaman et al., 2019a).

378 **TABLE 2** | Physicochemical characteristics of dairy waste effluents* (Danalewich et
379 al., 1998; Omil et al., 2003; Demirel et al., 2005; Byrne, 2011; Carvalho et al., 2013;
380 Karadag et al., 2015; Kolev Slavov, 2017; Verma and Singh, 2017; Goli et al., 2019;
381 Ferreira et al., 2021; Sivaprakasam and Balaji, 2021).

382 **TABLE 3** | Physico-chemical characteristics of DPS and cattle manures. The values
383 are collected from peer reviewed articles where data were given in form of range, mean or
384 median values. (1: Arun and Sivashanmugam, 2018; 2: Ashekuzzaman et al., 2019a; 3:
385 Ashekuzzaman et al., 2021b; 4: Carta-Escobar et al., 2004; 5: Carvalho et al., 2013;
386 Demirel et al., 2005; 7: Ferreira et al., 2021; 8: Frac et al., 2012; 9: Gayathri et al.,
387 2015; 10: Gogoi et al., 2021; 11: Sommer et al., 2013; 12: Rani et al., 2012; 13: Yadav
388 et al., 2009).

389

390 In France, information and data pertaining to the agronomic benefits and risks of applying organic
391 waste to agricultural soils was collated by Houot et al. (2014) and used to re-evaluate the EU

392 Sewage Sludge Directive (86/278) (CEC, 1986). This work emphasised the need for sludge used
393 in agriculture to have P recycling as a main priority, and that this use must not be a risk to the
394 environment or to human health due to their contents of heavy metals, organic trace compounds,
395 pathogenic microorganisms and pharmaceutical compounds. To avoid some of these concerns, the
396 EU Council Directive 86/278/EEC set limits for the content of heavy metals (Cd, Cu, Hg, Ni, Pb
397 and Zn) (CEC, 1986), and individual European countries have set limits for synthetic organic
398 compounds and pathogens (Hudcová et al., 2019).

399 In a comprehensive study across nine Irish dairy plants, the concentration of heavy metals (i.e. Cr,
400 Cu, Ni, Pb and Zn) was examined in all major DPS types with lowest concentrations found in DAF
401 sludge and highest in AD sludge (Ashekuzzaman et al., 2019a). Overall, the heavy metal
402 concentrations across all tested DPS samples were significantly lower than limits set by the EU
403 for avoiding accumulation in agricultural soil to which sludge is applied (CEC, 2008) and the
404 levels were below those of livestock manure (Sommer et al., 2013), and composts (Bernal et al.,
405 2017). The results of the Irish study are in line with the current knowledge on heavy metals content
406 of DPS (Table 4) and indicates that heavy metal concentrations will not be a limiting factor for
407 legal and safe application rate of DPS to agricultural soils (Ashekuzzaman et al., 2019a; Shi et al.,
408 2021b). The concentration varies between dairies and this is due to the diversity of the milk bio-
409 products and the various possible steps in the treatment of the effluent. It is important to have
410 knowledge pertaining to the heavy metal content of DPS and DPS-derived STRUBIAS products
411 before land application, because farmers and society must be assured that the heavy metal content
412 is lower (in soil and plants) than the limits given for use before making final decisions and rules
413 of use of the waste as a fertilizer (Shi et al., 2021b).

414 **TABLE 4** | Concentration (mg kg⁻¹ dry weight) of heavy metals in DPS, comparison
415 with European Union (EU) regulation upper limit values for sewage sludge (SS) and
416 a range of organic fertilizers. The values are collected from peer reviewed articles where
417 data were given in as ranges, mean or median values (1: Directive 86/278/EEC (CEC, 1986);
418 2: López-Mosquera et al., 2000; 3: Yadav et al., 2009; 4: Fraç et al., 2012; 5: Kumar et al.,
419 2008; 6: Fraç et al., 2017; 7: Ashekuzzaman et al., 2019a, 8: EBC, 2012; 9: Peyton et al.,
420 2016).

421

422 Organic Trace Compounds (OTCs) are chemical products (hydrocarbons and their derivatives,
 423 degradation products, solvents, etc.) present in organic waste or derived due to degradation of the
 424 organic compounds by the microorganisms in sewage treatment plants or in the soil. They often
 425 accumulate by biomagnification and bioaccumulation in biological organisms and cause
 426 irreversible damage to biological systems. They are directly or indirectly toxic to humans and
 427 animals (such as endocrine disruption and tumour initiation) (Barret et al., 2012). Table 5 presents
 428 the European Commission limit values for organic contaminants (CEC, 2009).

429 **TABLE 5** | Threshold values of organic contaminants in organic wastes that may be recycled to
 430 soil for crop production (option 2-3 of the 2009 European report, CEC (2009)).

Element or compound		Threshold value mg kg ⁻¹ DM
11 PAHs ¹	ace, phe, fluo, fluor, pyr, B(b,j,k)F, BaP, BghiP, indenoP	6
7 PCBs ²	28, 52, 101, 118, 138, 153, 180	0.8-0.8
PCDD/F ³	-	No limit / 100*
LAS ⁴	4	No limit / 5000
DEHP ⁵	-	No limit / 100
NPE ⁶	NP, NP1EO, NP2EO	No limit / 450

431 ¹PAH Polycyclic aromatic hydrocarbons; ²Polychlorinated biphenyls, ³PCDD/F: Polychlorinated dibenzo-
 432 p-dioxins and -furans; ⁴LAS: Linear alkylbenzene sulfonate; ⁵DEHP: Phthalate Di(2-ethylhexyl)phthalate;
 433 ⁶NPE: Nonylphenol-mono-ethoxylate. *PCDD/F in ng I-TEQ kg⁻¹ of dry matter

434

435 In Europe, there are proposals for limits on polycyclic aromatic hydrocarbons (PAHs) contents in
 436 municipal sewage sludge applied to land (CEC, 2010a,b). Depending on the country-specific
 437 regulations, the type of OTCs and the threshold limits differ. Where organic waste is land applied,
 438 the following three PAHs i.e. fluoranthene 5, benzo(b)fluoranthene 2.5, benzo(a)pyrene 2 and 7
 439 PolyChloroBiphenyls (PCBs) (PCB 28, 52, 101, 118, 138, 153, 180) are considered good
 440 indicators of compound resistant to biodegradation, and according to French regulation must be
 441 below concentration limits (Hudcová et al., 2019). In contrast, German authorities do not regulate
 442 PAHs, while the threshold for the sum of nine PAHs is more stringent in Denmark than for the

443 sum of three PAHs in France. Companies that handle wastes must be aware that the concentration
444 of 16 PAHs depends on the type of sludge (Boruszko, 2017). Research has shown that anaerobic
445 fermentation and post-flotation may reduce the content of PAHs up to seven times its original
446 concentration (e.g. 689 $\mu\text{g kg}^{-1}$ of DM) (Boruszko, 2017), with most of the reduction of
447 hydrocarbons taking place in the final phase of fermentation. The concentrations of PAHs in DPS
448 are low and do not exceed the amount allowed by the European Commission (CEC, 2009; Table
449 5). Therefore, their use in agriculture will not be limited by PAHs (Pérez et al., 2001; Boruszko,
450 2017), but it is recommended that these are investigated in DPS-derived STRUBIAS products.

451 DPS contains living microorganisms originating from the treated wastewater. There are pathogens
452 (viruses, bacteria, fungi and parasites) derived from animal manure (Sobsey et al., 2006), and if
453 dairy cows ingest grass from fields where dairy manure has been applied, then there is a risk for
454 transfer of pathogens in infectious levels to the cows and subsequently to milk and therefore to
455 wastes. The European Commission (CEC, 2000) has in its third draft of the Working Document
456 on Sludge, proposed the following thresholds for a range of bacteria and worms for sludge to be
457 recycled to soil: (1) *E. coli* less than 5×10^5 colony forming units (CFU) per gram (wet weight) of
458 conventional treated sludge; (2) for advanced sanitised sludge *E. coli* must be below 1×10^3 CFU
459 g^{-1} wet colony of treated sludge; (3) *Salmonella Senftenberg* W775 in sludge spiked with this
460 microorganism must be reduced 99.99%; (4) no content of *Ascaris ova* (5) a sample of 1 g DM of
461 the treated sludge must not contain more than 3×10^3 spores of *Clostridium perfringens*, and (6) a
462 sample of 50 g (wet weight) of the treated sludge must not contain *Salmonella spp.*

463 There are few studies reported about reduction of pathogens in raw DPS. Laboratory studies have
464 quantified reductions of microbial infectivity (inactivation) in animal organic wastes under
465 controlled temperature conditions and the samples have been stored aerobically or anaerobically
466 (Sobsey et al., 2006). For example, the high initial level of pathogens (*Enterobacteriaceae*, faecal
467 coli forms and *E. coli*) in dairy slurry was higher than in sludge from two urban wastewater
468 treatment plants where anaerobic digestion was followed by mechanical dehydration (one

469 treatment also received a heat-dried process). The concentrations in treated dairy slurry were 5.1
470 $\times 10^7$ CFU g^{-1} of DM of *Enterobacteriaceae*; 4.4 $\times 10^7$ CFU g^{-1} of DM of faecal coli forms, and 4
471 $\times 10^6$ CFU g^{-1} of DM of *E. coli* and all are below the limits set by the regulation. Concerning the
472 persistence of these pathogens in the soil, then after a 80-day trial across soil/sludge treatments,
473 the populations of faecal coliforms and *E. coli* decreased considerably or were not detectable
474 (Estrada et al., 2004). The study of Ravva et al. (2006) also found that the pathogenic strain *E. coli*
475 O157:H7 introduced in water from on farm dairy waste lagoons, failed to establish and proliferate
476 in dairy wastewater microcosms with or without circulating aerators. On the other hand, high
477 concentrations of *Listeria* are found in manure and sewage sludge and have survived in topsoil
478 between 12 and 182 days (Sobsey et al., 2006). If these are present in DPS, then additional sludge
479 treatments such as anaerobic digestion, hygienization by adding lime, or composting will reduce
480 the concentration of pathogens to allowable values.

481 Dairy processing wastes (even after treatment at source) may contain harmful substances, which
482 need testing and quantification across all the DPS and STRUBIAS types. The substances in focus
483 should be antimicrobial drugs, hormones, pesticides, emerging contaminants, pathogens,
484 disinfectants, persistent organic pollutant residues, microplastics and nanoparticles in DPS or DPS-
485 derived STRUBIAS (Shi et al., 2021a).

486

487 **DPS-derived STRUBIAS Production**

488 Another strategy that is being deployed to manage DPS is to further process these raw products
489 into other more usable and stable forms. Struvite (magnesium ammonium phosphate hexahydrate,
490 $MgNH_4PO_4 \cdot 6H_2O$) is widely used in agriculture due to its N and P content, which is in a form that
491 efficiently can be used by plants (Adam et al., 2009). Phosphorus can exist as particulate and
492 dissolved species in both organic and inorganic forms. The inorganic P species is mainly in
493 orthophosphate form, which is plant available and important for soil fertility but can be readily
494 lost to the environment (Frossard et al., 1996). However, the chemical composition of struvite
495 obtained from DPS is not always consistent with pure struvite equivalents (Hall et al., 2020). Metal
496 impurities such as Al, Fe, Ca and small amounts of heavy metals can precipitate along with the
497 struvite and could pose problems later for crops and soil when land applied.

498 The term “char-based materials” is used here to replace ‘biochar’ in the STRUBIAS acronym as
499 they have different terms depending on the technology. Char-based materials, obtained from the
500 thermochemical conversion of biomass in an oxygen-depleted atmosphere, are porous and
501 carbonaceous, and are more stable and C-rich and less toxic than the feedstock (Kambo and Dutta,
502 2015; Atallah et al., 2020;). The significance of thermochemical treatment lies in overcoming the
503 structural inferiority of biomass, which enhances the chances of energy and resource recovery from
504 waste (Kambo and Dutta, 2015). There are many functions of char-based materials including, but
505 not limited to, energy production, agriculture, C sequestration, wastewater treatment and bio-
506 refinery (Kambo and Dutta, 2015). The utility of a specific char-based material for any particular
507 application depends on its inherent properties. Feedstock, pre-treatment method, and temperature
508 are all important (Amoah-Antwi et al., 2020). However, thermochemical treatments increase the
509 risk of producing chars with other highly toxic compounds produced from high-temperature
510 reactions such as PAHs, PCBs, dioxins, furans, and PCDD/Fs (Kambo and Dutta, 2015; Amoah-
511 Antwi et al., 2020;). Heavy metals present in the feedstock are most likely to remain and
512 concentrate in the chars (Shackley et al., 2010).

513 Ashes are characterised as fly ash or bottom ash, or a combination formed through the incineration
514 of biowastes by oxidation (Huygens et al., 2019). Ash normally contains valuable plant
515 macronutrients such as K, P, S, Ca and Mg (Haraldsen et al., 2011; Knapp and Insam, 2011; Brod
516 et al., 2012). In addition, they contain large amounts of P (13.7%-25.7% P_2O_5), which are
517 comparable to commercial superphosphate (Xu et al., 2012). Obstacles to the use of ash as a
518 fertilizer or soil amendment could be their heavy metal contents (Franz, 2008; Herzel et al., 2016).

519

520 **DPS and DPS-Derived STRUBIAS as Fertilizers**

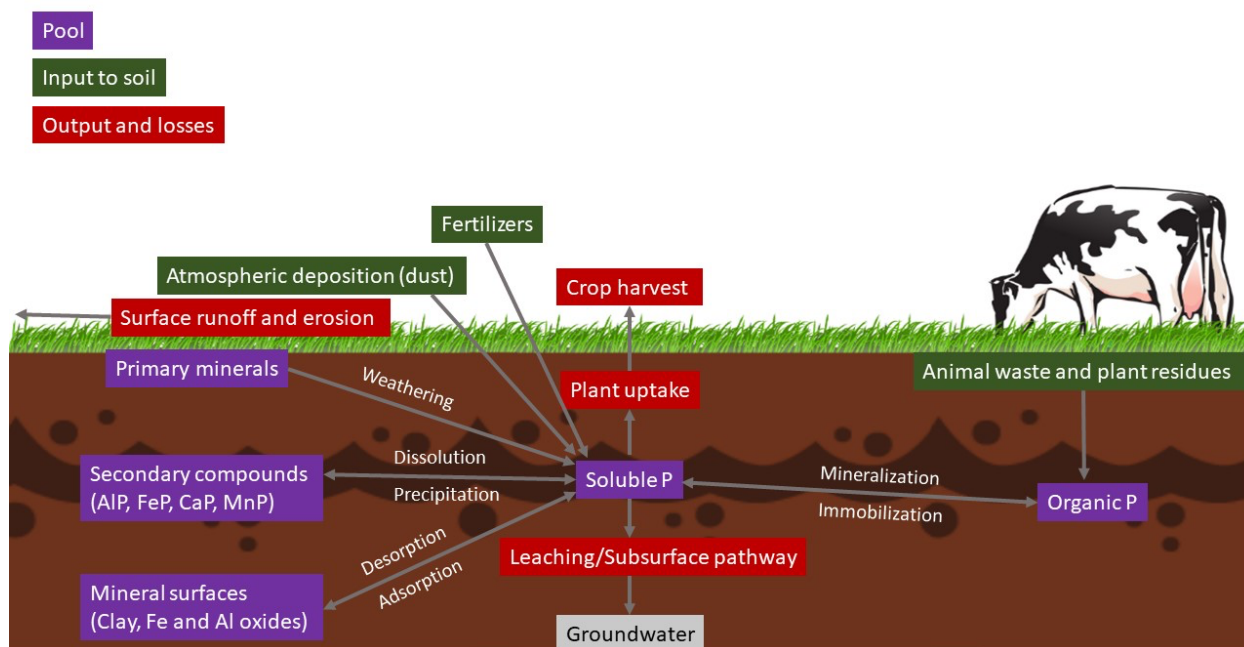
521 DPS and DPS-derived STRUBIAS products are used or research is underway to ascertain their
522 potential as bio-based fertilizers in agriculture (Shi et al., 2021a). DPS is typically stored off site
523 until applied to land in spring, whereas STRUBIAS products can be processed and stored until
524 needed. Many knowledge gaps still exist pertaining to their respective effects on P dynamics once

525 spread onto soils and their FEV. It should be noted that FEV is used herein (Shi et al., 2021a), but
526 can be often known in the literature as Mineral Fertilizer Equivalence (MFE; Delin, 2012) or
527 Mineral Fertilizer Replacement Value (MFRV; Schröder et al., 2007).

528

529 **Phosphorus Dynamics in Agricultural Soils**

530 In cropped agricultural systems, P applied to soil with fertilizers can be utilised by crops, absorbed
531 by soil minerals (González Jiménez et al., 2019), or lost along surface (runoff) or subsurface
532 (leaching and loss along natural or artificial lateral transport or deeper recharge to groundwater)
533 pathways to surface waters (Murnane et al., 2016). Soil P transformation passes through several
534 interconnected pools. These are the soluble P pool, which is considered to be immediately
535 available for plants; labile or weakly adsorbed P, insoluble P chemically bound with Ca ions in
536 calcareous and alkaline soils or occluded by Fe and Al oxides in acidic soils, P strongly adsorbed
537 by hydrous oxides of Fe and Al, and insoluble organic P within soil organic matter (Stevenson and
538 Cole, 1999; Bennett and Carpenter, 2002). Figure 3 presents a simplified diagram reflecting P soil
539 cycling and interactions between these pools. Briefly, the soil P cycle consists of the following
540 processes: weathering and precipitation, mineralization and immobilization, adsorption and
541 desorption, and P losses through surface or near surface runoff and subsurface leaching with
542 eventual recharge to groundwater (the proportions of which are dependent on soil/subsoil/bedrock
543 permeability and chemistry).



544

545 **Figure 3.** Soil phosphorus turnover in soil and pathways of P loss to waters on agricultural
 546 landscapes.

547

548 Mineralization and immobilization of P are part of the organic P cycle. Mineralization is a process
 549 of transformation of organic P to soluble H_2PO_4^- or HPO_4^{2-} . Mineralization of organic P slowly
 550 releases soluble P, which is crucial during the growing season as it provides a continuous supply
 551 of P to crops. Mineralization of organic P in soil occurs through breakdown of organic bonds,
 552 which is driven by the release of enzymes produced by plants and soil microflora. Organic P
 553 mineralization is driven by phosphatase enzyme activity in soil, which mainly occurs during the
 554 growing season when soil temperature ranges between 18 and 40°C (Prasad et al., 2016).
 555 Phosphatases synthesis are believed to be driven by P availability and enhances under P limiting
 556 conditions (Luo et al., 2017). However, in some cases application of mineral P (Paredes et al.,
 557 2011), and organic amendments of soil (Parham et al., 2002) can increase phosphatases activity.
 558 Some other factors which have an impact on phosphatases production are P availability and
 559 availability of other soil nutrients (Marklein and Houlton, 2012), soil moisture, pH, and availability

560 of other soil nutrients, and energy supply (Acosta-Martínez and Waldrip, 2014; Prasad et al.,
561 2016).

562 Precipitation and dissolution and desorption and absorption are part of inorganic P cycle. The
563 direction of P transfer between inorganic P soil pools through precipitation and dissolution can be
564 either reversible or irreversible, and can be impacted by a number of factors, including
565 geochemical soil composition and soil pH. For instance, in acidic soils P precipitation occurs in
566 the presence of Fe, Al, and Mg, and soluble P in such soil can be limited, while in alkaline soils
567 precipitation primarily occurs through reactions involving Ca^{2+} compounds (Prasad et al., 2014).
568 Adsorption, or fixation, binds soluble P compounds to soil particles, whereas desorption releases
569 P which is bound with soil minerals to soil solution, thereby increasing the soluble P pool. Unlike
570 precipitation, this process is reversible, and P does not involve permanent change in chemical and
571 structural changes in P-containing compounds and soil minerals.

572 The consideration of the aforementioned P fluxes in agricultural soil and recycled DPS
573 composition is essential for developing guidelines of alternative P fertilizers. Specifically products
574 derived from chemically treated dairy effluents treated with lime, ferric sulphate or aluminium
575 chloride may contain elements, which can limit P release into available P pool such as Ca, Fe, and
576 Al (Ashkuzzaman et al., 2019a). An inherent soil pH range optimal for P fertilizers to remain in
577 soluble pool is between 6 and 7.5. Decreasing soil pH can lead to soluble P fixation by Fe and Al
578 oxides. While fixation can be a limiting factor for soil P availability for crops (Daly et al., 2015;
579 Prasad et al., 2016), fixation of P by minerals present in the soil is also a limiting factor. To ensure
580 sustainable use of the P source and avoid P losses into the environment, such best practice should
581 be followed (Arenas Montaña et al., 2021).

582

583 **Fertilizer Equivalent Value of DPS and DPS-Derived STRUBIAS**

584 The FEV defined as the equivalent application rate of an inorganic fertilizer achieved by an organic
585 waste to achieve the same crop yield or nutrient uptake (Brod et al., 2012). The efficiency of most
586 bio-based organic fertilizers is lower than inorganic fertilizers because of their slow nutrient
587 release rates (Chen, 2006). The FEV of an organic fertilizer can both provide a quantitative
588 estimate of the amount of efficient nutrients in bio-based fertilizer and estimate of the actual value

589 when compared with a chemical equivalent. This information, which is currently lacking, would
 590 give growers accurate information to help with nutrient management planning on farms.

591 Two methods used to assess the FEV of bio-based fertilizers such as DPS or DPS-derived
 592 STRUBIAS products, are pot or field-scale studies, which include different fertilizer rates, crops
 593 and soils. The most common method is to compare yields or nutrient uptake results from DPS or
 594 DPS-derived STRUBIAS treatments with uptake from commercial mineral fertilizers as used with
 595 other organic fertilizers e.g. Lalor et al. (2011) examined the FEV of dairy cattle slurry. Typically,
 596 data fitted to linear, quadratic or cubic polynomial regressions, creates a relationship equation. For
 597 example, Figure 4 illustrates a fitted polynomial function, describing crop yield or nutrient uptake
 598 corresponding to different mineral fertilizer application rates. This is the method used to determine
 599 the corresponding mineral fertilizer rate (x_1) to any crop yield or nutrient uptake by a bio-based
 600 application. The mineral fertilizer rate, x_1 , expressed as a percentage of total nutrient applied from
 601 that bio-based treatment and estimates the FEV. Alternatively, calculation of FEV by the apparent
 602 nutrient recovery method without the need of a response curve is used. There is, however, a
 603 difference between apparent N or P recovery (ANR or APR) and N-P FEV. The first is the N or P
 604 fraction taken up by the test crop of total applied nutrients and the second is the ratio of the apparent
 605 N and P recovery of bio-based fertilizer and that of mineral fertilizer at the same rate (Cavalli et
 606 al., 2016, Sigurnjak et al., 2019). They are determined as follows using Equations 3-6:

607

$$608 \quad ANR = \frac{N \text{ uptake}_{TREATMENT} - N \text{ uptake}_{CONTROL}}{\text{total N applied}_{TREATMENT}} \quad (3)$$

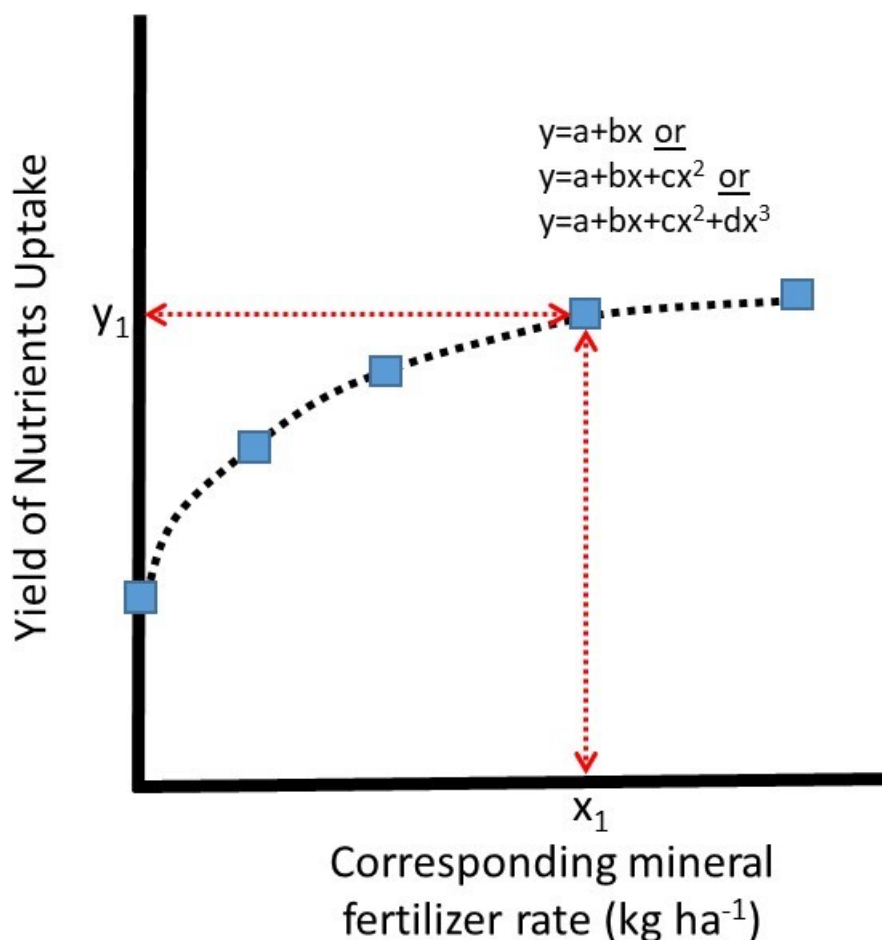
$$609 \quad NFEV(\%) = \frac{ANR_{DPS TREATMENT}}{ANR_{Mineral N TREATMENT}} \times 100 \quad (4)$$

$$610 \quad APR = \frac{P \text{ uptake}_{TREATMENT} - P \text{ uptake}_{CONTROL}}{\text{total P applied}_{TREATMENT}} \quad (5)$$

$$611 \quad PFEV(\%) = \frac{APR_{DPS TREATMENT}}{APR_{Mineral P TREATMENT}} \times 100 \quad (6)$$

612

613



614

615 **Figure 4.** Illustration of a FEV idealised response curve, where “a” is the intercept (crop yield or
 616 nutrients uptake at 0 kg ha⁻¹ of mineral fertilizer); “b, c and d” are the linear, quadratic and cubic
 617 coefficients, respectively.

618

619 The most comprehensive grassland study on the FEV of DPS, conducted by Ashekuzzaman et al.
 620 (2021a, b), examined two main types of DPS. The first is aluminium or iron-precipitated activated
 621 sludge (Al- or Fe-DPS) and the second is a lime-stabilised calcium-precipitated sludge (Ca-DPS).
 622 At field scale, an assessment of N and P availability for crop yield and uptake in comparison to
 623 reference mineral fertilizers over one seasonal year was undertaken. Ashekuzzaman et al. (2021b)
 624 found N-FEV of 22-25%, 54% and 8%, respectively, for Ca-DPS, Fe-DPS and Al-DPS. They
 625 indicated that N-FEV varied between activated and lime treated DPS types, as affected by
 626 wastewater and sludge treatment processes and storage. The different treatments affect the

627 proportion of mineral and organic N in the DPS, and thus the available N pool in amended soil.
628 With regards to P availability, the results of Ashekuzzaman et al. (2021a) show that first-year
629 cumulative P availability (over the four harvests) differs significantly between Al- and Ca-DPS
630 where Al-DPS P-FEV was 109% compared to mineral P (applied at 40 kg P ha⁻¹) and Ca-DPS P-
631 FEV was only 31%. Their findings show that mineral P fertilizer was a better starter fertilizer that
632 at application provided more readily available P for plant uptake than either Al-DPS or Ca-DPS,
633 as they observed 50% and 16% P-FEV for the two DPS respectively in the first harvest. Although
634 the Al concentration (1122 mmol kg⁻¹) in Al-DPS did not limit first-year P bioavailability, the
635 initial nature of P fractions, and their biological and bio-chemical mineralisation processes, might
636 be the reason of lower P availability for immediate uptake by plant. For Ca-DPS, high Ca content
637 (Ca/P molar ratio 1.86) and alkaline pH in Ca-DPS was likely to be associated with formation of
638 low soluble Ca-P compounds and low P availability. Future studies on the aspect of P composition
639 and mineralisation process in DPS would help to realize and correlate P uptake efficiency.

640 Other literature pertaining to the FEV of DPS, and especially DPS-derived STRUBIAS products
641 is still limited. Many factors affect the calculation of FEV such as the treatment processes used to
642 produce a DPS or DPS STUBIAS type (crop type, fertilizer application rate, duration of
643 experiment and scale of experiment (pot versus field) (Brod et al., 2012; Černý et al., 2012).

644 The dosing of Al or Fe salts used to capture P in sewage wastewater treatment plant affect P-FEV,
645 because high concentrations (more than 2800 mmol kg⁻¹) of either or both may significantly reduce
646 P bioavailability (Khiari et al., 2020). The P fertilizer effects of 14 different bio-based fertilizers
647 had been tested in pot experiments with ryegrass (Delin, 2016). At the first cut, the P-FEV of Fe-
648 and Al-precipitated sewage sludge were 37 and 33%, respectively (Delin, 2016). Falk Øgaard and
649 Brod (2016) found that P-FEV of 11 sewage sludges treated with Al and/or Fe salts varied
650 significantly between sludges, but was low for all sludges in a pot experiment with ryegrass. It was
651 lowest at first cut, where it ranged from 2 to 24% (Falk Øgaard and Brod, 2016). Both studies
652 indicated that sludge derived from a treatment with Fe had a higher P-FEV than when coagulation
653 with Al salts occurred. This is due to a higher solubility of Fe phosphate compared to Al phosphate.
654 The amount of Fe used is also important and sludge with a Fe:P ratio at 1:6 contains more plant
655 available P than sludge with a higher ratio (e.g. Fe:P ratio of 9:8) (Kahiluoto et al., 2015). Calcium
656 is another element that has an effect on the P availability. High dosing of Ca in the wastewater

657 with a Ca:P ratio of 2:1 reduce the P-FEV due to formation of Ca-P compounds such as
658 hydroxylapatite, which has a low solubility, an effect shown when using the sludge to produce
659 compost and biochar (Nest et al., 2021). As mentioned above, solubility of the P crystals are
660 affected by pH and liming increases the plant-available P in sludge produced from the wastewater
661 treated by Al or Fe salts (Krogstad et al., 2005; Montgomery et al., 2005; Bøen and Haraldsen,
662 2013).

663 When processing DPS to a STRUBIAS product, the FEV will change. This is influenced by the
664 untreated DPS physiochemical characteristics and the processing methods and parameters used.
665 For example, the P-FEV in ash produced from incineration is low when wood is used (P-FEV, 8-
666 16%), but gets higher using chicken manure (P-FEV = 13-39%) (Yusiharni et al., 2007), and
667 highest when incineration of biogas residue is used (MFE = 76-99%; Kuligowski et al., 2010). The
668 plant availability of P in thermochemical products such as ash and biochar depends on the
669 temperature during combustion/pyrolysis, and is halved by increasing the incineration temperature
670 from 400 to 700°C, which is due to hydroxyapatite formation (Thygesen et al., 2011). To increase
671 the amount of P in STRUBIAS products, the use of flocculants or biological processes to increase
672 P availability in raw DPS whilst avoiding high temperatures during the production of STRUBIAS
673 could be implemented.

674 For N-FEV, the proportion of ammonium N ($\text{NH}_4^+\text{-N}$) to the total N content and the C/N ratio of
675 the DPS or DPS-derived STRUBIAS products are the most important factors for the FEV (Sommer
676 et al., 2013; Webb et al., 2013). Ammonium is immediately available for the crop and is often a
677 growth-limiting factor (Brod et al., 2012; Gomez-Munoz et al., 2017). When fertilizer rates
678 increased, the N-FEV of meat and bone meal (MBM) and composted fish sludge (CFS) decreased
679 from 76 to 65% and 67 to 53%, respectively (Brod et al., 2012; Gómez-Muñoz et al., 2017). This
680 is consistent with crop response trials, where increasing amounts of N are applied (e.g. Brod et al.,
681 2012, see Figure 4). In that example, two industrial composts (i.e. neutral and acid Dynea
682 composts) had only N-FEV values ranging from 7 to 30%, as they contained low amounts of NH_4^+
683 and the N-mineralization rate was low. Acidification of the compost increased the ANR and N-
684 FEV of Dynea composts compared to untreated compost (Brod et al., 2012). This is due to a
685 reduced NH_3 emission during composting, as is seen when acidifying stored pig and cattle slurry.
686 In long-term studies adding human sewage sludge (i.e. biosolids) to silage maize, the FEV was

687 55% at low application rates and 64% at high application rates (Černý et al., 2012). This result,
688 compared with the finding by Brod et al. (2012), implied that in short-term fertilizer application,
689 doubling the application rate might not have a higher ANR and FEV, but in the long-term
690 application, an increase of the rate increase ANR and FEV. The reason can be that higher
691 application rate in short-term studies leads to emission of easily available N (NH₃ emission,
692 denitrification) and that reduces FEV, while in long-term studies there still is this immediate loss
693 but organic N increases in the soil and this will lead to higher amounts of N mineralised with time.
694 Gómez-Muñoz et al. (2017) found in the long-term experiment that continuous application of
695 agricultural and urban wastes improved soil quality, and long-term N availability correlates with
696 the accumulation of N and C in soil. That study reported the ANR and FEV in the final year (2013)
697 had generally increased compared to those in the first year of the study (2003), except for
698 composted household waste and cattle deep litter. The effect of C:N ratio was documented for
699 biochar produced by the pyrolysis of eucalyptus wood, as ANR values increased from 28%-40%
700 with increasing C:N ratios (2-4.9). Therefore, as new DPS-derived STRUBIAS products are
701 emerging, there needs to be a test phase before their use in agriculture. This should involve short
702 to long-term pot and field trials across crop and soil types to investigate P dynamics in soil and
703 their N-P FEV values.

704

705 **Potential Environmental Losses**

706 During the storage and land application of DPS, there may be the risk of nutrient loss or emissions
707 to waters (surface and subsurface) and/or the atmosphere, respectively.

708

709 ***Potential Losses From DPS/STRUBIAS to Waters***

710 As with all fertilizers, there is an associated risk of pollutants loss to waters (surface and subsurface
711 pathways) (Sørensen and Jensen, 2013) (Figure 3). DPS contains high levels of P and other
712 constituents such as C, N, Na and Cl that can alter soil composition and runoff behaviour (Liu and
713 Haynes, 2010 & 2011). The application timing and method of DPS are both important factors to

714 control to minimise pollutant losses to waters. Two recent studies have examined nutrient losses
715 from DPS in field soil experiments. The first micro-plot lab study applied several DPS types to a
716 grassland soil in Ireland and investigated the potential losses on P and N in runoff using simulated
717 overland flow after 48 h of DPS application (Ashekuzzaman et al., 2020). That study found that
718 the soluble P loss was highest for Ca-DPS (5.7 mg L^{-1}) followed by Al-DPS (0.8 mg L^{-1}) and Fe-
719 DPS (0.15 mg L^{-1}). In addition, P losses from DPS, including Ca-P-rich DPS, are much lower
720 when compared to cattle slurry (7.0 mg L^{-1}). With regard to N, that study observed dominant N
721 losses were $\text{NH}_4\text{-N}$ (nitrate (NO_3^-) losses were negligible) in the runoff pathway with
722 concentrations ranging from 2.6 and 3.3 mg L^{-1} . Such concentrations are significantly lower than
723 equivalent studies that focused on dairy cattle slurry (17.4 mg L^{-1}). The availability of N in organic
724 wastes can be predicted from their C:N ratio (Delin et al., 2012), and DPS has a C:N ratio of ~6
725 (Ashekuzzaman et al., 2019a), which is comparable with human sewage sludge. According to
726 Delin et al. (2012), this implies that around 50% of the N content is easily available, as also found
727 for sewage sludge (Petersen et al., 2003). The second field study examined P accumulation in soil
728 and potential losses in surface runoff and leaching (multi-depth) at seven sites in New Zealand
729 (Lizarralde et al., 2021). Results showed that after the long-term application of DPS (based on N
730 content), high amounts of P in the soil at least to 30-cm depths accumulated. The level of
731 accumulation varied across soils and was due to the history of wastewater application, the capacity
732 of the soils to sorb P and the land use and system management.

733 Organic fertilizers such as DPS delivered and applied on arable land (e.g. winter cereals) can be
734 an effective component of any nutrient management plan. For practical reasons, DPS is often
735 applied in autumn before sowing a winter cereal like winter wheat. However, under free draining
736 soils (loamy sand and sandy loam soils with a yearly drainage surplus of 300-400 mm) and wet
737 and cool North-European conditions, extra NO_3^- leaching losses, equivalent to 20-30% of total N,

738 can be expected after application of organic fertilizers with similar N availability to winter wheat
739 in autumn (Sørensen and Rubæk, 2012). Under conditions with less surplus precipitation or
740 application to crops with a large capacity for N uptake in autumn, less NO_3^- leaching by application
741 in autumn are expected. By waste application in spring, NO_3^- leaching is significantly lower
742 (Sørensen and Rubæk, 2012) and NO_3^- leaching is often proportional to total N application (De
743 Notaris et al., 2018; Pedersen et al., 2021) and thus nearly similar for organic N and mineral N.
744 The total N applied with organic fertilizers is higher than with mineral fertilizers to obtain the same
745 fertilizer value and thereby crop yield. This also implies that NO_3^- leaching is higher by application
746 of organic wastes in spring compared to mineral N fertilization, but higher leaching losses can be
747 prevented by use of cover crops (Pedersen et al., 2021). Pedersen et al. (2021) found extra NO_3^-
748 leaching equivalent to 8% of the N input in the first year and 4% in the second year after application
749 for both mineral and organic N applied to a loamy sand and a sandy loam soil in spring.

750 In contrast to N, soluble P is strongly bound to soil implying that very low leaching losses of P
751 occur after application of organic fertilizers. However, if DPS is applied directly to soil and not
752 incorporated or injected, there is risk of incidental P losses (0.3 – 7.6% of total input) by surface
753 runoff (Ashkuzzaman et al., 2020) and by leaching through macropores in soil (Sørensen and
754 Jensen, 2013). Such losses can both occur by transport in soluble form (e.g. dissolved reactive
755 phosphorus) and in the form of particle and colloid-bound P. Christiansen et al. (2020) found large
756 variation in water-extractable P (0.1 to 9 % of total P) in various sludge types and therefore the
757 risk of soluble P loss is also variable. A reduced risk of P losses along surface runoff and subsurface
758 macropore leaching pathways by incorporation of DPS into soil or by injection is possible
759 (Sørensen and Jensen, 2013). When soils are loaded with excessive amounts of P over a longer
760 period, the soil is saturated with P and P leaching to drains is significantly increased (Heckrath et
761 al., 1995). By precipitation of P and N in struvite, nutrients become concentrated like in mineral
762 fertilizers and can be stored and applied as for mineral equivalents. This means that its application
763 can occur following best practice for precision farming i.e. right time, right place, right amount,
764 right method and right product.

765 After pyrolysis of sludge for biochar production, most of the organic N is lost. The availability and
766 fate of this N is not well investigated e.g. Christiansen et al. (2020) found that a biochar derived
767 from a mixture of human sewage sludge and straw contained 5% of total P in water-extractable

768 form and most of the P content was soluble in a weak acid (citric acid). Therefore, a part of the P
769 in sludge-based biochar solubilises in soil. Weak biochar binding on clay minerals and its low
770 density can lead to environmental losses to waters. In addition, translocation of biochar due to
771 hydrological connectivity is observed. For instance, Rumpel et al. (2006) showed that biochar
772 accumulates at the bottom of slopes within the landscape and such losses are important to quantify
773 as they can be delivered to surface water (Major et al., 2010). Therefore, biochar needs to be
774 incorporated into soil to avoid loss of P in surface runoff either in soluble or in particulate form.
775 This is also the case in grasslands, where a significant reduction in P losses may occur where
776 injection rather than surface application of manure is practiced (Uusi-Kämppe and Heinonen-
777 Tanski, 2008). This precision farming application method where available could be a DPS
778 application method that minimises incidental losses of pollutants in runoff during rainfall events.

779

780 *Potential losses from DPS/STRUBIAS to the atmosphere*

781 To date, there have not been many studies that have measured or calculated the accumulated
782 emissions of GHG and NH₃ from the production, storage or land application of these products
783 (Figure 5). Therefore, DPS or STRUBIAS emissions of CH₄, N₂O and NH₃ from production until
784 after field application are calculated using a combination of information about emission from the
785 products or by using similar products as a proxy. Herein, such risks and mitigation for each
786 management step i.e. from production, to processing of the sludge, to storage of the sludge and
787 sludge products, and to application of DPS and secondary STRUBIAS products is considered.

788 The DPS is usually stored anaerobically until application to soil. During this phase CH₄ and NH₃
789 may be emitted; little N₂O is emitted during the storage phase as the waste tends to not have a
790 surface crust where nitrification-denitrification may take place (Baral et al., 2018). However,
791 application to soil emits N₂O (Scott et al., 2000; Yoshida et al., 2015). A fraction of the C in sludge
792 applied to soil will contribute to C storage. The emitted NH₃ can contribute to N₂O emission after
793 deposition to land or water. Transforming sludge into biochar, hydrochar or ash will cause an
794 emission of CO₂, but this treatment will eliminate CH₄ emission and may affect N₂O emission
795 when applied to soil. The CO₂ emitted during treatment of the sludge is part of the circulation of

796 C between the atmosphere, plants, intake by dairy cows and recycling of the waste and therefore
797 considered climate warming neutral.

798 In recent field studies, biochar has not increased N₂O emissions from “fertilized soil” when applied
799 to fields (Clough and Condon 2010; Taghizadeh-Toosi et al., 2011; Liao et al., 2020; Thers et al.,
800 2020). There is no emission of N₂O during storage of biochar, as it has been shown in compost
801 studies that biochar reduces N₂O production and emission (Shakoor et al., 2021). In several meta-
802 analyses it has been shown that N₂O emissions from soil decreases after the addition of biochar
803 (Cayuela et al., 2014; Sri Shalini et al., 2020). Reasons for reduced N₂O emission from soil treated
804 with biochar could be improved soil aeration, increased soil pH, enhanced N immobilization, and
805 possible toxic effect induced by biochar organic compounds (polycyclic aromatic hydrocarbons)
806 on nitrifier and denitrifier communities (Taghizadeh-Toosi et al., 2011; Cayuela et al., 2014; Harter
807 et al., 2014). In contrast, some studies show increased N₂O emission from soil with biochar, which
808 is attributed to an increased soil water content in the presence of biochar favouring denitrification,
809 or the release of biochar embodied-N (Lorenz and Lal, 2014).

810 The N content of biochar or hydrochar is not high due to the transformation of N during initial
811 feedstock thermolysis (Majumder et al., 2019). Although N content in biochar or hydrochar is low,
812 the application of biochar materials into the soils can affect the soil N cycle. Biochar and hydrochar
813 have been shown to adsorb NH₄⁺ on biochar particles and reduce NH₃ volatilization; however, the
814 increased NH₃ volatilization, observed from some soil treated with hydrochar, is possibly due to
815 the reduced ability to absorb NH₄⁺ associated with greater hydrophobicity of hydrochar (Clough
816 and Condon, 2010; Taghizadeh-Toosi et al., 2012a; Subedi et al., 2015).

817 Carbon in DPS and in DPS-derived STRUBIAS products added to soil will contribute to soil C
818 storage, the sequestering potential or mean residence time (MRT) being related to the rate of
819 transformation of the added carbon to CO₂ (Tian et al., 2009). It has been shown that CO₂ fluxes
820 were suppressed when biochar was added to fertilized soils (Wang et al., 2016), which may be due
821 to reduced enzymatic activity and the precipitation of CO₂ onto the biochar surface (Case et al.,
822 2014). Ethylene, which is frequently present in biochar, can sometimes inhibit the transformation
823 of C in soil (Spokas et al., 2010). However, if there is labile C input in biochar or hydrochar, it can
824 result in positive priming effects (He et al., 2017) and increased CO₂ emissions, although part of

825 the CO₂ may have originated from carbonate formed during pyrolysis (Kuzyakov et al., 2009).
 826 Pyrolysis and gasification materials have been assessed by many (e.g. Lal, 2009; Beesley et al.,
 827 2011; Wu et al., 2017) to increase soil organic C content and to improve overall soil health.

828

829 **Figure 5.** GHG emission from scenario of dairy sludge management no treatment of sludge,
 830 pyrolysis and hydrothermal (HTC) treatment of sludge.

831

832 The present literature review shows that STRUBIAS production potentially can reduce GHG
 833 emission from all sites of the sludge management chain. An analysis of the emission of GHG from
 834 sludge is stored until it was applied to soil or alternatively processed, chars stored and then applied
 835 to soil was carried out to provide insight in the potential total GHG reduction due to production of
 836 biochar and hydrochar (Figure 5). In calculations carried out using the model outlined in material
 837 and methods, the total GHG increase in the atmosphere due to sludge managed traditionally is 359
 838 kg CO_{2eqv} (Table 6). In contrast to sludge management, CH₄ emission during storage of
 839 STRUBIAS products is avoided, and N₂O emission from biochar is negligible and from hydrochar
 840 reduced to 1/3 of the emission from sludge, because 55% of the N in sludge is recycled to the
 841 wastewater plant. Due to the recalcitrant nature of C in STRUBIAS products, more C is
 842 sequestered when these are applied to fields. Within the boundary of the sludge managing system,
 843 the emission from standard sludge management is 359 kg CO_{2eqv} (an increase in CO₂ in the
 844 atmosphere). Producing hydrochar reduces GHG in the atmosphere corresponding to -30 kg CO_{2eqv}
 845 and biochar production to -92 kg CO_{2eqv}. Avoiding CH₄ and N₂O emission from the sludge are the
 846 most important factors to reduce the climate warming potential of dairy waste management.

847 **TABLE 6** | Greenhouse gas emission from the management chain of dairy sludge
 848 management untreated and after HTC or pyrolysis – calculated in the present study.
 849 Sludge composition is dry matter (DM) 170 g kg⁻¹, Ammonium-N 0.94% in DM,
 850 Total-N 35% in DM, Total C 39.0% in DM. Negative numbers means reduction in
 851 GHG in the atmosphere and positive increase GHG concentrations.

Source	Dairy sludge	Hydrobiochar	Biochar
	kg CO ₂ ekv.		
Stored sludge, CH ₄	247		-

Nitrous oxide, N ₂ O	182	55	-
Carbon sequestration	-70	-85	-92
Reduction of GHG	359	-30	-92

852

853

854 **Effects on Soil Microorganisms**

855 Soil microorganisms play a pivotal role in nutrient cycling in agricultural ecosystems. Their
 856 activities enhance the availability of essential nutrients for crop growth and contributes to an
 857 improvement of soil properties such as OM content and water retention capacity. The application
 858 of organic residues like DPS modifies soil microbial communities significantly. These
 859 modifications are highly variable and depend strongly on the composition of the bio-based residues
 860 applied.

861

862 *Microbial Analysis*

863 The methods used to assess soil microbial processes divide into abundance, diversity and activities.
 864 Determination of soil microbial biomass in soil by chloroform fumigation is a common indicator
 865 applied after the application of different types of sludge (Charlton et al., 2016). Other techniques
 866 that can provide information on microbial biomass are phospholipid fatty acid assays (PLFAs) and
 867 the quantification of total DNA. These two techniques are also powerful methods to determine
 868 changes in the diversity of soil microorganisms. DNA-based techniques such as 16S and 18S gene
 869 quantification, metagenomics and metabarcoding, provide very insightful information on the
 870 community composition (Abdelfattah et al., 2018; Bünemann et al., 2018; Bastida et al., 2019).
 871 PLFAs allow distinguishing between bacteria and fungi, and further distinctions between bacterial
 872 groups such as Gram⁺ and Gram⁻ bacteria (Frostegård and Bååth, 1996). Other commonly used
 873 methods to determine changes in microbial metabolic diversity are community-level physiological
 874 profiling (CLPP) assays such as Biolog Ecoplates (Liu et al., 2017). The latter technique provides
 875 profiles of potential degradation of different complex chemical C substrates, which assess the
 876 ability of soil microbial communities to degrade natural soil constituents (Siebielec et al., 2018).
 877 Lastly, enzymatic activities are the most common techniques to study soil microbial activities after

878 the application of organic residues. The most common enzymes measured are phosphatases, β -
 879 glucosidases, dehydrogenases and ureases (Table 7). Enzymes are substrate-specific and can be
 880 associated with different nutrient cycles in the soil (Burns et al., 2013).

881 **TABLE 7** | Information on relative changes of microbial indicators after the application of
 882 sewage and dairy sludge to soil. Relative differences were calculated by comparing the results
 883 from controls (unfertilized or mineral fertilizer). Results shown are the results of the most
 884 distinctive differences between the treatments and controls. Variation is only showed when
 885 significant results are indicated. References: 1: Fraç and Jezińska-Tys 2011; 2: Oszust et al.,
 886 2015; 3: Fraç 2012; 4: Jezińska-Tys and Fraç, 2009; 5: Gryta et al., 2014; 6: Brookes and
 887 McGrath 1984; 7: Charlton et al., 2016; 8: Banerjee et al., 1997; 9: Fliesbach et al., 1994; 10:
 888 Abaye et al., 2005; 11: Roy et al., 2019; 12: Fernandez et al., 2009; 13: Jorge-Mardomingo et al.,
 889 2013; 14: Torsvik et al., 1998; 15: Dar 1996; 16: Houben et al., 2019; 17: Bastida et al., 2019;
 890 18: Nicolás et al., 2014; 19: Kızılkaya and Bayraklı 2005; 20: Kunito et al., 2001; 21: Markowicz
 891 et al., 2021; 22: Siebelec et al., 2018; 23: Liu et al., 2017; 24: Singh et al., 2014.

Sludge type	Analysis	Analysis	Percentage of variation	Ref.
Dairy sludge	Enzymatic activities	Acid phosphatase*	85	1
Dairy sludge	Enzymatic activities	Dehydrogenase	26.32 to 2500	2, 3, 4
Dairy sludge	Enzymatic activities	Urease	77.78 to 750	1, 4
Dairy sludge	Enzymatic activities	Protease	+250 to 3150	1, 4
Dairy sludge	Microbial diversity	CLPP ¹	-61.54 to 160	2, 3, 5
Sewage sludge	Microbial biomass	Microbial C	-64.46 to 250	6, 7, 8, 9, 10, 11, 12, 13, 14, 15
Sewage sludge	Microbial biomass	Microbial P	+85.71	16
Sewage sludge	Microbial biomass	PLFA ²	18.95 to 50.45	10, 17, 18

Sewage sludge	Enzymatic activities	Alkaline phosphatase	-66.67 to 129.62	11, 15, 19, 20, 21, 22
Sewage sludge	Enzymatic activities	Acid phosphatase*	-15.49 to 400	11, 13, 21, 22
Sewage sludge	Enzymatic activities	Dehydrogenase	-82.19 to 600	11, 13, 15, 22, 21
Sewage sludge	Enzymatic activities	β -Glucosidase	-60 to 1000	13, 19, 20
Sewage sludge	Enzymatic activities	Arylsulphatase	-72.97 to 141.67	19, 20
Sewage sludge	Enzymatic activities	Urease	-50 to 50	13, 19, 21
Sewage sludge	Enzymatic activities	Catalase	+266.67	13
Sewage sludge	Enzymatic activities	Protease	-50 to 350	13, 20
Sewage sludge	Microbial diversity	Bacterial population	23.28 to 764.91	10, 11, 17, 18, 22
Sewage sludge	Microbial diversity	Fungal population	-31.17 to 250	11, 18, 22
Sewage sludge	Microbial diversity	CLPP ²	-66.1 to 28.57	8, 21, 23
Sewage sludge	Microbial diversity	16S ³	-10 to 5	21, 24

892 *= Phosphatase measured at pH 6.5; †= Comparison between metal-contaminated dairy sludge; ¹:
893 CLPP : Community level physiological profiling; ²: Phospholipid fatty acid assays; ³: 16S: bacterial
894 rDNA

895

896 ***Effects of DPS Application on Soil Microbial Communities***

897 Little information is available and from a small number of research teams on the direct effects of
898 dairy sludge on soil biological properties, and the studies do not always come to the same
899 conclusion (Table 7). Reported increases of microbiological indexes and improvements in soil
900 nutrient cycling after the application of DPS appear in the literature. Soil enzymatic activities (Frac

901 and Jezierska-Tys 2011; Fraç et al., 2012; Oszust et al., 2015) and soil microbial diversity as
902 revealed by CLPPs analyses (Fraç et al., 2012; Oszust et al., 2015) were reported to increase after
903 the application of dairy sludge. Gryta et al. (2014) show that the effect of heavy metal
904 contamination due to DPS application caused a significantly lower (-61.54 %) C degradation as
905 shown by a Biolog Ecoplate assay and DPS is a source of heavy metal contamination in soils
906 (López-Mosquera et al., 2000). Heavy metal concentrations below guideline values found in
907 studies by Ashekuzzaman et al. (2020) and Shi et al. (2021a) point to the importance of regulated
908 application rates (based on P) but also recognising heavy metal application which varies DPS
909 types. As wastewater treatment is improved, heavy metal concentrations should be reduced further
910 e.g. biological P, removal strives to replace the need for metal coagulants.

911 Similar results were found when assessing other types of sludge such as sewage sludge, which has
912 a high content of C and contributes to increased soil microbial biomass and activities (Torsvik et
913 al., 1998; Abaye et al., 2005). Yet, similar to the results of Gryta et al. (2014) on dairy sludge,
914 sewage sludge effects on soil microorganisms depend also on the amount of pollutants. Sewage
915 sludge may contain large concentrations of pollutants such as heavy metals and pathogens that
916 often can decrease soil microbial indicators (Torsvik et al., 1998; Charlton et al., 2016; Major et
917 al., 2020) (Table7). Although concentrations of heavy metals and other pollutants may be low in
918 DPS compared to sewage sludge, there is a need for long-term field trials, especially pertaining to
919 bioaccumulation of heavy metals in soil (and crops) as this has the potential to damage soil
920 microbial communities.

921 Gryta et al. (2014) examined DPS heavy metal content on soil biology. Their results showed that
922 high concentrations of heavy metals led to a decrease in most soil biological descriptive variables,
923 associated with damage of the cell membrane, mitochondria or DNA (Dar, 1996; Tchounwou et
924 al., 2012). Torsvik et al. (1998) compared the application of contaminated sewage sludge with
925 unpolluted equivalents to examine microbial diversity. Using a technique that uses “total number
926 of genomes”, as equivalent to the *E. coli* genome, their results indicated that the application of
927 polluted sludge had negative impacts on soil biodiversity, causing a reduction of up to 6.5 times
928 less biodiversity. Previous experiments also compared polluted versus unpolluted sewage sludge
929 with similar conclusions (Brookes and McGrath, 1984; Fliesbach et al., 1994). Other studies using
930 Biolog (CLPPs) assessments (Banerjee et al., 1997) and PLFA analyses have shown similar results

931 (Singh et al., 2014) (Table 7). Similarly, several studies show the depressed activities of soil
932 microorganisms by reduced enzymatic activities after polluted sludge applications to soil (Dar
933 1996; Kunito et al., 2001; Kızılkaya and Bayraklı, 2005; Speir et al., 2007; Fernández et al., 2009;
934 Markowicz et al., 2021) (Table 7).

935 Another risk derived from the application of DPS is the introduction of pathogens and organic
936 pollutants. Research from sewage sludge shows the presence of human and animal pathogens such
937 as *E. coli*, *Listeria*, *Clostridium perfringens*, *Enterococcus* or *Salmonella* (Brochier et al., 2012).
938 These pathogens may survive on plant tissues, in soils and in hydroponic systems to which sludge
939 is applied (Brochier et al., 2012; Kyere et al., 2019). Native soil microbial communities are known
940 to decrease the survival of potential pathogens (Xing et al., 2020), but specific strains of bacteria,
941 such as *E. coli* O104:H4, may survive in soil for over a year after its inoculation (Knödler et al.,
942 2016). Moreover, sewage and DPS introduce antibiotic resistant genes to soils, causing the
943 development of antibiotic resistant bacteria with severe implications on human and environmental
944 health (Rizzo et al., 2013; Dungan et al., 2018; Urra et al., 2019). Nevertheless, even if pathogens
945 are present in DPS, their concentration is believed to be 10 to 15 times lower than in sewage sludge
946 (Kwapinska et al., 2020). As a consequence, the risks of introducing pathogens and other organic
947 pollutants after the application of DPS are smaller than other residues such as sewage sludge.

948 Study results with respect to the effect of sludge applications in agriculture and their effects on
949 soil communities vary, but for dairy derived sludge most studies show increases in microbial
950 functional diversity (CLPP) and activities (Table 7). Activities of dehydrogenase (Jeziarska-Tys
951 and Frąć, 2009; Frąć et al., 2012; Oszust et al., 2015), acid phosphatase (Frąć and Jeziarska-Tys,
952 2011), urease and protease (Frąć et al., 2012; Oszust et al., 2015) have been reported to increase
953 after the application of dairy sludge (Table 7). CLPP experiments have also revealed increased
954 degradation of C substrates (Frąć et al., 2012; Oszust et al., 2015).

955 When sewage sludge is not heavily polluted, similar trends can be observed (Table 7). Enzymatic
956 activities have been shown to increase with the application of sewage sludge (Dar, 1996; Jorge-
957 Mardomingo et al., 2013; Siebielec et al., 2018; Roy et al., 2019), soil microbial biomass (Charlton
958 et al., 2016) and diversity indexes has been reported to improve based on Biolog assessments (Liu

959 et al., 2017). DNA approaches have also revealed little effect on soil microbial diversity and
960 bacterial antibiotic-resistance after the application of sludge (Rutgersson et al., 2020).

961 The application of unpolluted sludge to soils might be very beneficial in improving soil health and
962 fertility. Sludge contains large concentrations of easily decomposable C, readily available N, P
963 and other essential nutrients for plant and microbial growth (Krogstad et al., 2005; Singh and
964 Agrawal, 2008; Peltre et al., 2011). The recalcitrant components of sludge release at a slow rate
965 by specific microbial groups such as P solubilising microorganisms (PSM) (Clarholm, 1985; Khan
966 et al., 2009; Kuypers et al., 2018). The relative distribution of the different nutrient pools in sewage
967 sludge is highly dependent on the production, treatments and origin of the sludge (Singh and
968 Agrawal 2008).

969 The high C content constitutes the most significant attribute of sludge that can affect the
970 development and growth of soil microbial communities, because C is the most limiting element
971 for bacterial and fungal growth in soils (Demoling et al., 2007; Hobbie and Hobbie, 2013).
972 Boosting the C content in soil leads to a concatenated stimulus in the cycling of other nutrients
973 such as N or P (Demoling et al., 2007). Field studies have also confirmed the positive effects of
974 sludge application on both N and P cycles (Hallin et al., 2009; Fraç and Jezierska-Tys 2011;
975 Houben et al., 2019). The application of large amounts of C is associated with a significant growth
976 in soil microbial biomass (Charlton et al., 2016; Houben et al., 2019). Yet, this positive effect of
977 sewage sludge on soil microbial communities has been reported to lead to significant nutrient
978 immobilization by soil biota (Smith and Tibbett 2004; Gómez-Muñoz et al., 2017). It is expected
979 that sewage sludge with a high C:N ratio (>15) might lead to N immobilization by soil biota
980 (Gómez-Muñoz et al., 2017). The same applies for the immobilization of P, application of organic
981 materials with a high C:P ratio might lead to its immobilization (Zhang et al., 2018). Whenever
982 sludge is applied as a bio-based fertilizer, these aspects should be considered. However, in the long
983 term, the application of C-rich materials such as sewage sludge should improve soils from an
984 agronomic and environmental point of view. Building up C content in soils would improve nutrient
985 cycling, providing a slower release maintained over time and lower losses that might contaminate
986 soils and water bodies (Gómez-Muñoz et al., 2017; Zhang et al., 2018).

987

988 **4. Conclusion**

989 This review collated information that will help give DPS and their secondary products certification
990 as P-fertilizers in accordance with technical proposals for new fertilizing materials under
991 forthcoming EU Fertilizing Product Regulations. It presents the current state of knowledge
992 pertaining to dairy processing sludge and STRUBIAS bio-based fertilizers and identifies
993 knowledge gaps and potential solutions to minimise environmental losses to soil, water and air.

994 STRUBIAS products have a high P concentration compared to that of sludge, and a well-defined
995 fertilizer efficiency of the P applied to fields. To achieve high P fertilizer efficiency, dairy
996 wastewater treatment must aim at producing sludge with soluble P-components and avoid Al
997 coagulation and P insolubility. In STRUBIAS production, conditions producing less soluble P
998 should be avoided, i.e. high temperatures. The benefits of STRUBIAS production is a reduction
999 of transport cost of P due to a high P concentration. In the development of production units, it is
1000 important that STRUBIAS products can be applied with traditional mineral fertilizer application
1001 machinery. Heavy metal concentration of known products are below the limits set for the use of
1002 these as fertilizers, and the risk of disease spreading and negative effects on microbial activity in
1003 soil is low. Producing STRUBIAS products eliminates GHG emissions from management of the
1004 sludge from dairies. A goal of STRUBIAS production could be recycling of plant nutrients and C
1005 to organic farms, thereby providing a sustainable circular economy. More information is needed
1006 to carry out an economic analysis (e.g. a cost comparison across dairy sludge management,
1007 production and management systems for STRUBIAS products and mineral fertilizer), which
1008 should include a value chain analysis of the whole system.

1009

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1013

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